

A FACILE ACCESS TO MASKED ISOBENZOFURANS ; HIGH EXO-STEREOSELECTIVITY  
IN THE DIELS-ALDER REACTIONS OF 4,7-DIHYDRO-4,7-ETHANOISOBENZOFURAN<sup>1</sup>.

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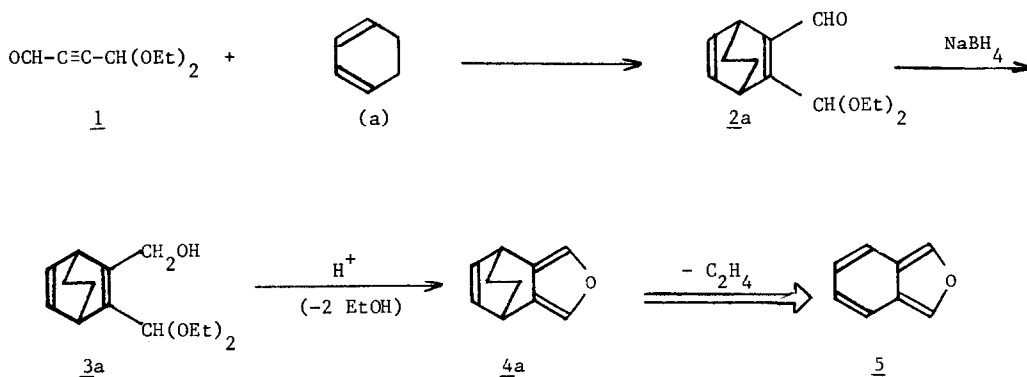
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Summary- The isobenzofuran derivatives precursors 4a-c are readily prepared from 1 ; their Diels-Alder adducts with cis-dienophiles present a stereochemistry endo>exo from 4b,c and, interestingly, only exo (towards the furan moiety) from 4a.

Owing to their great reactivity, isobenzofuran (IBF) and its derivatives<sup>2,3</sup> have been used, of late, as intermediates in the synthesis of polycyclic compounds including biomolecules such as anthracyclines and lignans. Besides reverse Diels-Alder processes, their generation has been achieved by elimination of HOR from 1-alkoxyphthalans<sup>4,5</sup> and, more recently, by cyclisation of  $\gamma$ -hydroxyacetals.<sup>6-8</sup> We have defined, on these grounds, a new simple route<sup>9</sup> to 4,7-dihydroisobenzofuran derivatives 4 from the readily available monoacetal of acetylenedicarbaldehyde 1.<sup>10,11</sup> We have also studied the Diels-Alder reactivity of the compounds with usual cis-dienophiles in order to compare the stereochemistry of the adducts produced to those of IBF itself, given that the latter generally affords a mixture of diastereoisomers with endo > exo.

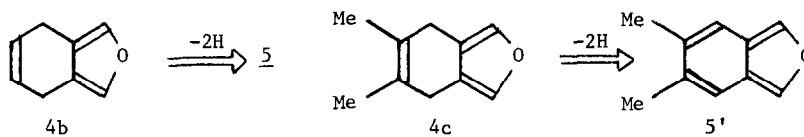
Thus 2a<sup>12</sup> (90 % improved yield from 1 and 1,3-cyclohexadiene (a), 3 days at 75°C under inert atmosphere) is cleanly reduced (NaBH<sub>4</sub>, MeOH, 95% yield) to the corresponding  $\gamma$ -hydroxyacetal 3a. (scheme 1). When small amounts of acid (HCO<sub>2</sub>H or HCl) are added to a



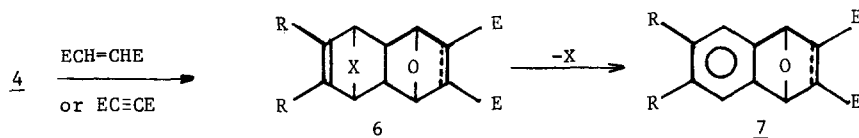
SCHEME 1

$\text{CH}_2\text{Cl}_2$  solution of the latter, a loss of ethanol occurs very readily giving rise to 4a, isolated (75% yield), after aqueous work-up and  $\text{SiO}_2$  column chromatography (pentane-ether 95:5), as a colourless solid, m.p.  $25^\circ\text{C}$ ,  $^1\text{H}$  n.m.r. ( $\text{CDCl}_3$  with  $\text{Me}_4\text{Si}$ )  $\delta$  1.60 (m, 4H, bridge), 3.80 (m, 2H, bridgehead), 6.52 (m, 2H, CH=) and 7.15 (s, 2H, =CH-O).

Similarly reaction of 1 with 1,3-butadiene (b) and 2,3-dimethyl-1,3-butadiene (c) afford compounds 4b<sup>13</sup> and 4c<sup>11</sup> respectively (overall yields 60 % and 68% from 2b and 2c).



These furans can act as convenient precursors of the corresponding IBF derivatives 5 and 5' since they react with ethylenic or acetylenic dienophiles (dimethyl maleate and fumarate, maleic anhydride, N-phenylmaleimide, methyl vinyl ketone, dimethyl acetylenedicarboxylate) to afford the expected cycloadducts 6 (or directly 7a depending on the thermal conditions), which could be aromatized either on heating (loss of  $\text{C}_2\text{H}_4$ ) for 6a or by DDQ oxidation<sup>14</sup> for 6b,c into 7 (scheme 2).



4a :  $\text{X}=\text{CH}_2-\text{CH}_2$ ,  $\text{R}=\text{H}$  ; 4b :  $\text{X}=\text{2H}$ ,  $\text{R}=\text{H}$  ; 4c :  $\text{X}=\text{2H}$ ,  $\text{R}=\text{Me}$ .

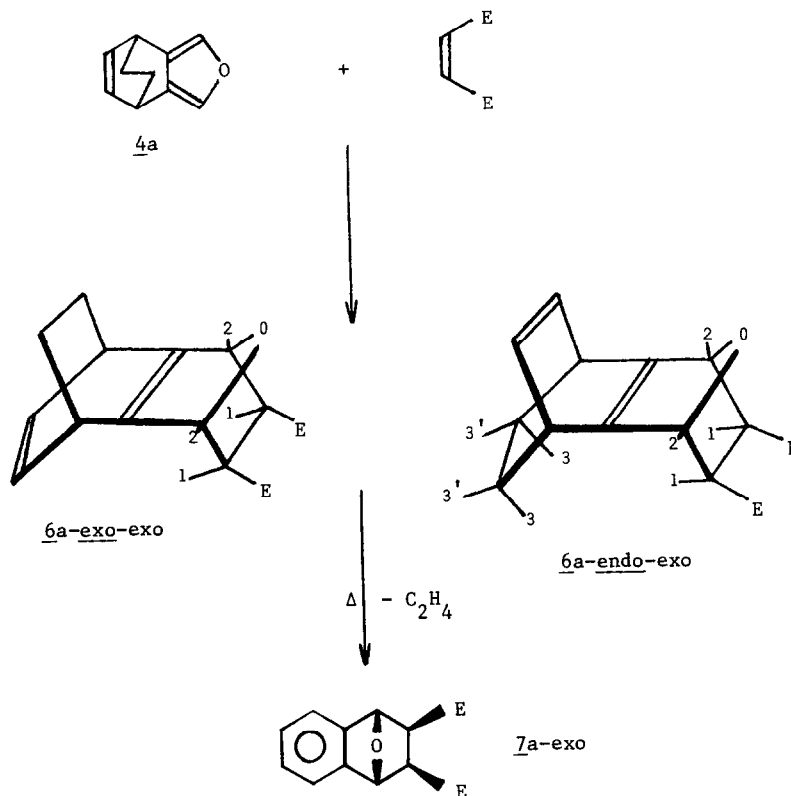
#### SCHEME 2

Whereas the stereochemistry of the Diels-Alder reaction of 4b,c with cis-dienophiles resembles that of IBF itself,<sup>2</sup> with formation of a mixture of endo and exo isomers of 6b,c or 7b,c (endo > exo, 2-3:1), interestingly, only exo additions take place with 4a.

For instance, after refluxing a  $\text{CHCl}_3$  solution of 4a for 20 hrs in the presence of an excess of dimethyl maleate (8 mol per mol), the exo adduct 7a only (scheme 3) is produced (70% isolated yield) as a white solid, m.p.  $150^\circ\text{C}$ , litt.<sup>15</sup>  $146-7^\circ\text{C}$ .

The cycloaddition can proceed at lower temperature with more potent dienophiles such as maleic anhydride and N-phenylmaleimide enabling the isolation of the non-aromatized cycloadducts 6a. For example, from N-phenylmaleimide (2 mol per mol of 4a,  $\text{CH}_2\text{Cl}_2$  solution, 2.5 hrs,  $20^\circ\text{C}$ ), 6a is obtained (70% yield) as a white solid, m.p.  $138-156^\circ\text{C}$ , consisting of unequal amounts of the two diastereoisomers exo-exo (75%) and endo-exo (25%), these assignments being established by  $^1\text{H}$  n.m.r. : no coupling is observed between  $\text{H}_1$  and  $\text{H}_2$  in both isomers (exo configuration), and nuclear Overhauser experiments (at 300 MHz) give a  $\Delta$  intensity ( $\text{H}_1$ ) = + 16% by irradiation of the ethylene protons  $\text{H}_3$  only in the minor isomer (endo configuration). On heating this mixture at  $160^\circ\text{C}$  for 5 mn it is converted quantitatively into the exo isomer of 7a, white solid, m.p. (benzene)  $208^\circ\text{C}$ . Since none of the other

endo isomers of 6a could be detected by monitoring this cycloaddition at low temperature ( $^1\text{H}$  n.m.r.,  $-20^\circ\text{C}$ ,  $\text{CDCl}_3$ ), we assume that the reaction proceeded under kinetic control through a sterically favoured exo approach of the dienophile onto 4a.<sup>16</sup>



SCHEME 3

In conclusion, the present methodology for the synthesis of 4 constitutes a good new entry into IBF chemistry with interesting stereochemical implications in the case of 4a. Its generalisation to the case of isoindole precursors will be later reported.

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